



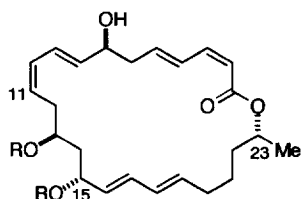
Enantioselective Synthesis of the C11-C24 Segment of Macrolactin A via Organoiron Methodology

Vadapalli Prahlad and William A. Donaldson*

Department of Chemistry, Marquette University, P.O. Box 1881, Milwaukee, WI 53201-1881

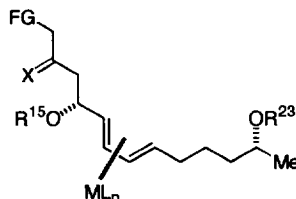
Abstract: The enantioselective synthesis of the $\text{Fe}(\text{CO})_3$ complexed C11-C24 segment of macrolactin A has been accomplished from *rac*-(methyl 6-oxo-2,4-hexadienoate) $\text{Fe}(\text{CO})_3$ in 11 steps (>50% ee).
Copyright © 1996 Elsevier Science Ltd

Macrolactin A (**1a**) is a 24-membered polyene macrolide isolated from a taxonomically-undefined deep sea bacterium.^{1a} The structure of **1a** was assigned on the basis of NMR spectroscopy,^{1a} and by chemical degradation and synthesis of the fragments.^{1b} This compound exhibits antiviral activity against Herpes Simplex I and II and against HIV. Unfortunately, the culturing of this bacterium has been "unreliable".^{1b} The unique structure of **1a** coupled with its enticing biological activity and relative scarcity from natural sources has led to synthetic studies by several research groups.² Boyce and Pattenden have recently reported preparation of 13,15-dimethoxymacrolactin A (**1b**) using Stille Pd-coupling methodology for construction of the diene linkages.³ The C11-C24 segment **2b** was a key intermediate in their synthesis. We here report on the preparation of a similar C11-C24 segment **2a** which relies on the *stereodirecting ability of an $\text{Fe}(\text{CO})_3$ adjunct to control introduction of the remote asymmetric centers at C15 and C23*.



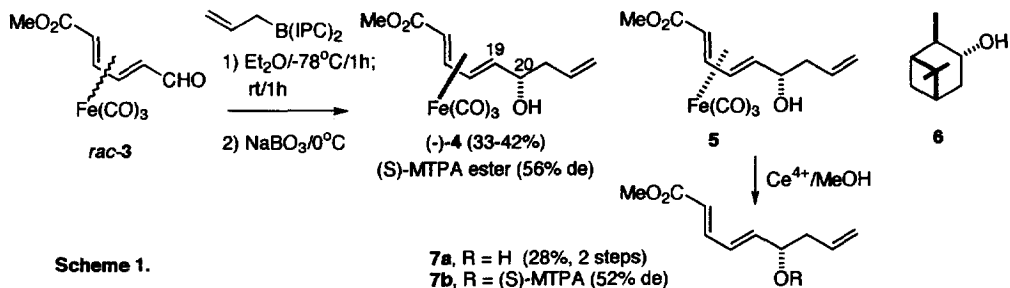
1a, R = H

1b, R = Me

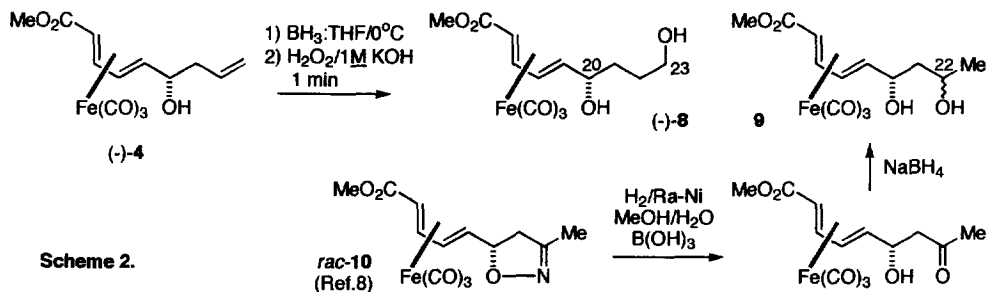


| | FG | X | R ¹⁵ | R ²³ | ML _n |
|-----------|-----------------|--------|-----------------|-----------------|--------------------------|
| 2a | 2-(1,3-dioxane) | O | H | OTBDMS | $\text{Fe}(\text{CO})_3$ |
| 2b | CHO | OMe, H | Me | OAc | none |

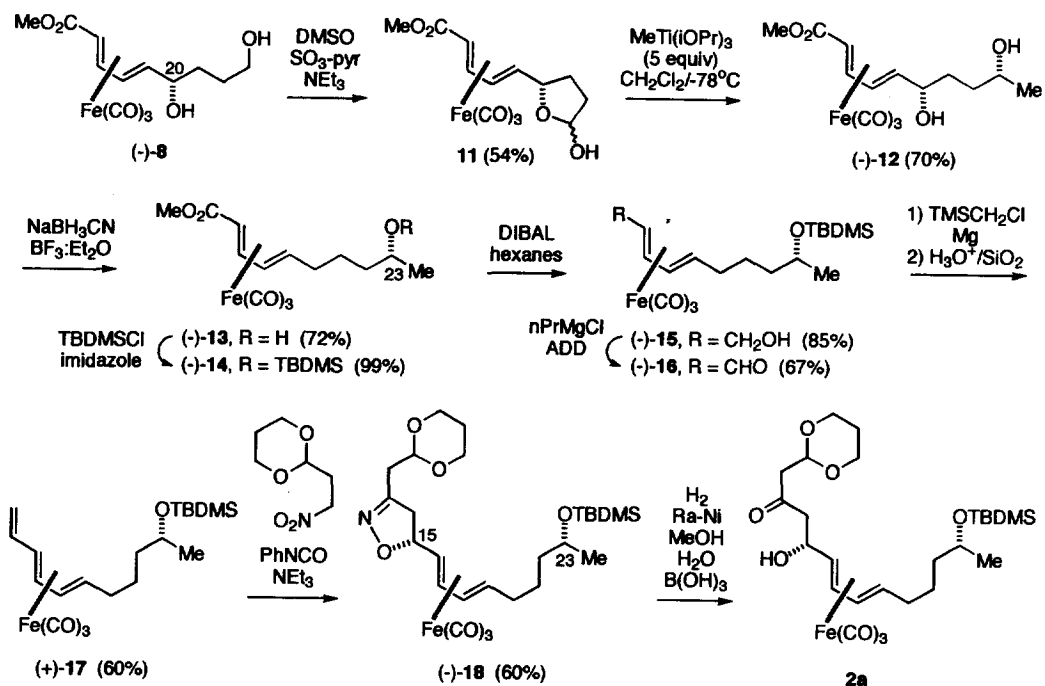
The reaction of (methyl 6-oxo-2,4-hexadienoate) $\text{Fe}(\text{CO})_3$ (*rac*-**3**)⁴ with *B*-allyldiisopinocampheylborane (prepared from (-)-(IPC)₂BOMe),⁵ followed by brief oxidative work up with NaBO_3 , gave a mixture of alcohols (-)-**4**, **5** and isopinocampheol **6** (Scheme 1). After chromatographic separation, the desired (-)-**4** was assigned the 19*S*,20*S* stereochemistry (54% ee, macrolactin A numbering) by comparison of its optical rotation with that of the previously reported⁶ *R,R* enantiomer. The (*S*)-Mosher's ester of (-)-**4** was determined to be 56% de by ¹H NMR spectroscopy. While (-)-**4** was readily separable, the diastereomeric alcohol **5** could not be easily separated from **6**. However, treatment of a mixture of **5** and **6** with $(\text{NH}_4)_2\text{Ce}(\text{NO}_3)_6$ gave the free ligand **7a**, which was easily separable from **6**. The derived (*S*)-MTPA ester **7b** was determined to be 52% de by ¹H NMR spectroscopy.



The hydroboration-oxidation of **4** has been previously reported⁷ to afford only the 1,4-diol **8** (91%). In our hands, treatment of (-)-**4** by the literature procedure gave a separable mixture of (-)-**8** (57%) and 1,3-diol diastereomers **9** (35%). The identity of **9** was established by comparison to a racemic sample prepared by independent synthesis (Scheme 2). Reductive hydrolysis of the known⁸ isoxazoline *rac*-**10**, followed by reduction of the resultant hydroxyketone gave *rac*-**9** as a mixture of C22 epimers. The formation of significant amounts of product resulting from Markovnikov addition in the hydroboration of 4-hydroxy-1-alkenes has been previously noted.⁹ The 23-TBDPS ether-20-(S)-MTPA ester of (-)-**8** was determined to be 57% de by ¹H NMR spectroscopy.



While oxidation of (-)-**8** with PDC or QDC occurred predominantly at the C20 pseudo-benzylic hydroxyl group, it was found that Moffat oxidation gave predominantly **11** as a mixture of lactol epimers (54%, Scheme 3).¹⁰ Reaction of **11** with MeTi(iPrO)₃¹¹ gave a *single* diastereomeric diol, (-)-**12** (70%),¹² which was tentatively assigned the syn-1,4 stereochemistry on the basis of literature precedent¹¹ and by analogy to our previous model compound studies.^{2c} Selective ionic reduction^{2b,c,13} of the C20 alcohol group of (-)-**12** gave the known^{2b} (-)-**13** (72% yield, 53% ee by $[\alpha]_D$).¹⁴ The absolute stereochemistry of (-)-**13** (i.e. 23R) was assigned on the basis of the relative chemical shifts of the C24 methyl groups of the corresponding (R)- and (S)-MTPA esters (δ 1.35 and 1.27 ppm respectively).¹⁵ Conversion of (-)-**13** to the aldehyde (-)-**16** (TBDMSCl/imidazole/ DMAP, 99%; DIBAL/hexanes, 85%; nPrMgCl/1,1'-(azodicarbonyl)dipiperidine, 67%) followed the method of Grée, *et al.*^{2b} The enantiomeric excess of (-)-**16** was assayed by treatment with (1S,2S)-N,N'-dimethyl-1,2-diphenylethylenediamine/molecular sieves to generate the diastereomeric imidazolidines.¹⁶ Integration of the diastereomeric methyl groups of the crude product (δ 2.54 and 2.19 ppm vs δ 2.35 and 2.25 ppm) indicated these to be 55% de. Peterson methenylation of (-)-**16** gave the complexed triene (+)-**17** (60%).



Scheme 3.

With successful installation of the C23 stereocenter relative to the (diene)Fe(CO)₃ group, attention was focused on introduction of the C15 stereocenter. It was anticipated, on the basis of literature precedent,⁸ that nitrile-oxide cycloaddition to (+)-**17** would occur on the *s-trans* rotamer on the face opposite to the bulky tricarbonyl iron adjunct. In the event, reaction of (+)-**17** with 2-(2'-nitroethyl)-1,3-dioxane in the presence of phenylisocyanate and triethylamine led to the isolation of (-)-**18** (Scheme 3). The relative stereochemistry of isoxazoline (-)-**18** was assigned as Ψ-*exo* by comparison of its ¹H NMR spectral data¹⁷ to that of a model compound of known relative stereochemistry. In particular, for (-)-**18**, the signals for the C15 methine proton and the diastereotopic C14 protons appear at δ 4.26 (q), 3.18 (dd) and 2.86 (dd) ppm respectively, while the corresponding signals for the Ψ-*exo* [3-methyl-5-(1,3-pentadienyl)-isoxazoline]Fe(CO)₃^{8,13} appear at δ 4.24 (q), 3.09 (dd) and 2.74 (dd) ppm respectively. In this fashion, the relative stereochemistries at two centers (C15 and C23) which are 9 carbons separated, was controlled by use of the Fe(CO)₃ adjunct. Reductive hydrolysis of (-)-**18** (H₂, Ra-Ni, MeOH/H₂O, B(OH)₃, 5h) gave the β-hydroxy ketone **2a** in modest yields.¹⁸ Further processing of **2a** was precluded due to a dearth of material, however it is anticipated that the desired C13,C15 anti-diol stereochemistry could be established by reduction of **2** with Me₄NBH(OAc)₃.^{2f,19}

In summary, the C11-C24 segment of macrolactin A has been prepared in an optically active form using organoiron methodology. The (diene)Fe(CO)₃ functionality controls both C23 stereocenter via a chirality relay strategy and the C15 stereocenter via a highly diastereoselective nitrile oxide-olefin cycloaddition.

Acknowledgment. Financial support for this work was provided by the National Institutes of Health (GM-42641). High resolution mass-spectral determinations were made at the Nebraska Center for Mass Spectrometry and the Washington University Resource for Biomedical and Bio-organic Mass Spectrometry. The authors thank Ms. Zhi Wang for the preparation of a racemic sample of **11**, Mr. Bireshwar Dasgupta for preparation of 2-(2'-nitroethyl)-1,3-dioxane, and Dr. René Grée for generously providing spectral data and optical rotations for **13-16**.

References and Notes

- (1) (a) Gustafson, K.; Roman, M.; Fenical, W. *J. Am. Chem. Soc.* **1989**, *111*, 7519-24; (b) Rychnovsky, S.D.; Skalitzy, D.J.; Pathirana, C.; Jensen, P.R.; Fenical, W. *J. Am. Chem. Soc.* **1992**, *114*, 671-7.
- (2) (a) Rychnovsky, S. D.; Pickering, D. A. *Abstracts of papers*, 207th National Meeting of the American Chemical Society, San Diego, American Chemical Society: Washington, DC, **1994**, ORGN 209; (b) Benvegno, T.; Schio, L.; Le Floc'h, Y.; Grée, R. *Synlett*, **1994**, 505-6; (c) Donaldson, W. A.; Bell, P. T.; Wang, Z.; Bennett, D. W. *Tetrahedron Lett.*, **1994**, *35*, 5829-32; (d) Tanimori, S.; Morita, Y.; Tsubota, M.; Nakayama, M. *Syn. Commun.*, **1996**, *26*, 559-67; (e) Benvegno, T.; Toupet, L. J.; Grée, R. *Tetrahedron*, **1996**, *52*, 11811-20; (f) Benvegno, T.; Grée, R. *Tetrahedron*, **1996**, *52*, 11821-26.
- (3) Boyce, R. J.; Pattenden, G. *Tetrahedron Lett.*, **1996**, *37*, 3501-4.
- (4) Grée, R.; Tourbah, H.; Carrie, R. *Tetrahedron Lett.* **1986**, *27*, 4983-6.
- (5) Brown, H. C.; Jadhav, P. K. *J. Am. Chem. Soc.* **1983**, *105*, 2092-3.
- (6) Roush, W. R.; Park, J. C. *Tetrahedron Lett.* **1990**, *31*, 4707-10.
- (7) Grée, D.; Grée, R.; Lowinger, T. B.; Martelli, J.; Negri, J. T.; Paquette, L. A. *J. Am. Chem. Soc.* **1992**, *114*, 8841-6.
- (8) Le Gall, T.; Lellouche, J.-P.; Toupet, L.; Beaucourt, J.-P. *Tetrahedron Lett.* **1989**, *30*, 6517-20.
- (9) Brown, H. C.; Unni, M. K. *J. Am. Chem. Soc.* **1968**, *90*, 2902-5.
- (10) **11** (mixture of lactol epimers): IR (KBr, cm^{-1}) 3374, 2066, 1973, 1715; ^1H NMR (400 MHz, CDCl_3) δ 5.85 (dd, $J = 5.1, 8.1$ Hz, 1H), 5.58 (m, 0.5H), 5.48 (t, $J = 4.1$ Hz, 0.5H), 5.43 (m, 1H), 4.03 (q, $J = 7.8$ Hz, 0.5H), 3.84 (q, $J = 8.3$ Hz, 0.5H), 3.67 (s, 3H), 2.56 (br s, 1H), 2.3-1.7 (m, 4H), 1.33 (t, $J = 8.3$ Hz, 0.5H), 1.17 (m, 1H), 1.10 (d, $J = 8.7$ Hz, 0.5H); ^{13}C NMR (CDCl_3 , diastereomeric signals in parentheses) δ 172.4 (172.3), 99.4 (99.2), 86.1 (86.0), 85.0 (84.8), 82.8 (80.0), 65.5 (63.0), 51.7, 46.6 (46.5), 34.5 (33.6), 32.1 (31.3); HRMS (EI) m/z 338.0079 (calcd for $\text{C}_{13}\text{H}_{14}\text{O}_7\text{Fe}$ (M^+) 338.0093); Anal. Calcd for $\text{C}_{13}\text{H}_{14}\text{O}_7\text{Fe}$: C, 46.18; H, 4.17. Found: C, 46.42; H, 4.17.
- (11) Tomooka, K.; Okinaga, T.; Suzuki, K.; Tsuchihashi, G. *Tetrahedron Lett.*, **1987**, *28*, 6335-8.
- (12) (-)-**12**: mp 108-111°C; $[\alpha]_{\text{D}}^{-110}$ (c 0.16, MeOH); ^1H NMR (CDCl_3) δ 5.83 (dd, $J = 5.1, 8.1$ Hz, 1H), 5.50 (dd, $J = 5.1, 8.7$ Hz, 1H), 3.88 (m, 1H), 3.66 (s, 3H), 3.57 (m, 1H), 1.92-1.82 (m, 1H), 1.76-1.50 (m, 5H), 1.33 (t, $J = 7.8$ Hz, 1H), 1.23 (d, $J = 6.3$ Hz, 1H), 1.04 (d, $J = 8.1$ Hz, 1H); ^{13}C NMR (300 MHz, CDCl_3) δ 172.6, 85.5, 83.9, 73.4, 68.2, 67.2, 51.7, 46.0, 35.5, 35.3, 23.7; HRMS (FAB) m/z 355.0468 (calcd for $\text{C}_{14}\text{H}_{19}\text{O}_7\text{Fe}$ ($\text{M}+\text{H}$)⁺ 355.0480); Anal. Calcd for $\text{C}_{14}\text{H}_{19}\text{O}_7\text{Fe}$: C, 48.04; H, 5.25. Found: C, 47.48; H, 5.12.
- (13) Bell, P. T.; Dasgupta, B.; Donaldson, W. A. *J. Organomet. Chem.* submitted for publication.
- (14) (-)-**13**: $[\alpha]_{\text{D}}^{-120}$ (c 0.16, MeOH); ^1H NMR (300 MHz, CDCl_3) δ 5.78 (dd, $J = 5.1, 8.1$ Hz, 1H), 5.22 (dd, $J = 5.1, 8.7$ Hz, 1H), 3.80 (m, 1H), 3.65 (s, 3H), 1.76-1.30 (m, 7H), 1.20 (d, $J = 6.3$ Hz, 3H), 0.97 (d, $J = 8.4$ Hz, 1H); ^{13}C NMR (CDCl_3) δ 172.6, 87.2, 83.0, 67.7, 65.3, 51.6, 45.6, 38.6, 34.1, 28.2, 23.6. This NMR spectral data matches that kindly provided by Dr. René Grée.
- (15) Dale, J. A.; Mosher, H. S. *J. Am. Chem. Soc.* **1973**, *95*, 512-9.
- (16) Mangeny, P.; Alexakis, A.; Normant, J. F. *Tetrahedron Lett.* **1988**, *29*, 2677-80.
- (17) (-)-**18**: $[\alpha]_{\text{D}}^{-36}$ (c 0.16, MeOH); IR (CDCl_3 , cm^{-1}) 2047, 1979; ^1H NMR (300 MHz, CDCl_3) δ 5.23 (dd, $J = 4.9, 8.2$ Hz, 1H), 5.08 (dd, $J = 5.0, 8.8$ Hz, 1H), 4.74 (t, $J = 4.8$ Hz, 1H), 4.26 (q, $J = 9.3$ Hz, 1H), 4.11 (m, 3H), 3.76 (m, 4H), 3.18 (dd, $J = 10.3, 17.6$ Hz, 1H), 2.86 (dd, $J = 8.6, 17.6$ Hz, 1H), 2.64 (t, $J = 3.9$ Hz, 2H), 2.07 (m, 1H), 1.73-1.20 (m, 7H), 1.10 (d, $J = 6.0$ Hz, 3H), 1.02 (t, $J = 8.9$ Hz, 1H), 0.88 (s, 9H), 0.04 (s, 6H); ^{13}C NMR (CDCl_3) δ 155.0, 99.4, 86.3, 83.5, 83.0, 68.2, 66.8, 65.8, 59.6, 45.0, 39.1, 34.3, 33.6, 28.2, 25.8, 25.4, 23.7, 18.0, -4.5, -4.8; HRMS (FAB) m/z 564.2072 (calcd for $\text{C}_{26}\text{H}_{42}\text{O}_7\text{NSiFe}$ ($\text{M}+\text{H}$)⁺ 564.2079).
- (18) Attempted reductive hydrolysis of (-)-**18** over 48 h resulted in loss of the $\text{Fe}(\text{CO})_3$ adjunct, reduction of the conjugated diene, and reduction of the ketone functionality.
- (19) Evans, D. A.; Chapman, K. T.; Carreira, E. M. *J. Am. Chem. Soc.* **1988**, *110*, 3560-78.